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Key

Reviewing Key

Concepts 1. (a)

Sample answer:

Bend the spoon,
break it in

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pieces, and melt it. (b) Lighting a match is not a physical change.

2. (a) If a new substance is formed, a chemical change has occurred.

(b) In electrolysis, water is broken down into hydrogen and

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oxygen, two
different
substances.

Answer Key Ch

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...

Sample answer:
An element is a
pure substance
that cannot be
broken down into

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any other
substances. A
compound is a
pure substance
made of two or
more elements
chemically
combined in a
set ratio. 25.

Sample answer:
According to the
law of
conservation of
mass, matter is

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not created or
destroyed in any
chemical or
physical change.

6

Chapter 1 Test – Introduction to Matter

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instance,
formation of a
precipitate, gas
production, and
the color change
are possible
evidence that a
chemical
reaction has
taken place.

CH 5 Lesson 1
(Observing
Chemical Change)

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Recognizing

Chemical Changes

(page 54) 7.

List the four possible clues to a chemical change? 8. Is the following statement true or false? If you observe a clue for chemical

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Chemical Changes
Answer Key Ch
change, you can
be certain that
a chemical
change has taken
place. _____ 9.

Define a
precipitate.

Conservation of
Mass (page 55)
10.

SECTION 2.1
PROPERTIES OF
MATTER (pages

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39–42)

Chemistry 12th
Edition Chang,
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2.4 Chemical
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Objectives
Explain how
chemical
reactions affect
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Describe how energy changes affect how easily a chemical reaction will occur. Explain why enzymes are important to living things.

Lesson Summary
Chemical Reactions
Everything that

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happens in an
organism is
based on
chemical
reactions.

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Rar

a number in
front of a
chemical formula
in an equation
that indicates
how many
molecules or

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atoms of each reactant and product are involved in a reaction.

synthesis a chemical reaction in which two or more simple substances combine to form a new, more complex

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