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The infra-red  
spectra of H, I and J  
are given below,  
but it does indicate  
which is - which.

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Identify the three compounds H, I and J, using the infra-red spectra below, and decide which spectrum belongs to which compound. (iii) Compound E, which is a branched chain haloalkane, was found to have the composition by

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Answers  
mass of 39.8% C,  
7.3% H, 2)

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1) 2) 3) 4) 5) 6)

bromoethane —¥

ethyl ethanoate

cold, aq NaOH

nucleophilic

substitution

propene 2-methylp

ropylamine

chloroethane

ethanoic acid

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PROBLEMS

2-bromo-2-methylb

utane 4 5-40 (CH<sub>3</sub>

CH<sub>2</sub>), 5-40 (CH<sub>3</sub>

CH<sub>2</sub>), 10-70 (CBr),

5-40 ((CH<sub>3</sub>)<sub>2</sub> CBr)

methylpropene ) 3

90-150 (CH<sub>2</sub>), 90

-150 (C<sub>5</sub> H<sub>3</sub> )

propene 3 ) 90-150

(CH<sub>2</sub>), 90 -150 (C<sub>5</sub>

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Answers

2-chloropropane ),  
2 5-40 (CH<sub>3</sub> 10-70  
(CHCl) propanone  
, 2 20-50 (CH<sub>3</sub>  
190-220 (CO)  
methylamine 1  
25-60 (CH<sub>3</sub>)

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2,2-dimethylamine  
2-methylbut-1-ene  
3-methylbutan-1-ol  
4-hydroxybutanone  
3-methylbutanoic  
acid butyl  
methanoate  
ethanenitrile  
propanal Decide  
which spectrum  
belongs to which  
compound and  
draw the molecule  
next to the



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spectrum.

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029 (Spectroscopy)

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infra-red spectra of H, I and J are given below, but it does indicate which is - which. Identify the three compounds H, I and J, using the infra-red spectra below, Page 4/25

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Propene reacts with HBr to form H. H reacts with sodium hydroxide to form I, and I reacts with warm acidified potassium dichromate (VI) to form J. The infra-red spectra of H, I and J are given

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below, but it does indicate which is - which. Identify the three compounds H, I and J, using the infra-

H Nmr

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equations to give  
the molar ratio in  
which the following  
acids react with  
bases complete the  
table to show your  
answers acid  
formula of acid  
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1 deduce the rate equation for each of the following reactions a a reacts with b to form c a b c experiment, 3 1 5 kinetics questions by topic aqa ...

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IR PROBLEMS 1 The

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IR spectra of six compounds are shown. Complete the table to match the spectra to the compounds.

Identify any key signals you used to identify each compound. You may not be able to decide between two of the compounds.

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Structure Spectrum

Bond(s)

Wavenumber range

(cm<sup>-1</sup>) Structure

Spectrum Bond(s)

Wavenumber range

(cm<sup>-1</sup>) butanoic ...

IR PROBLEMS 1 The

IR spectra of six

compounds are

shown ...

rate =  $k[A][B]^2$   $k =$

rate = 16 = 2.0

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mol<sup>-2</sup> dm<sup>6</sup> s<sup>-1</sup>

[A][B]<sup>2</sup> (2 x 2.2) b)

$$\text{rate} = k[E] \quad k = \frac{\text{rate}}{[E]} \\ = \frac{0.80}{4} = 0.20 \text{ s}^{-1}$$

[E] 4 c) rate =

$$k[P][Q] \quad k = \frac{\text{rate}}{[P][Q]} \\ = \frac{0.00200}{0.1 \times 0.1} = 0.20$$

mol<sup>-1</sup> dm<sup>3</sup> s<sup>-1</sup>

[P][Q] 0.1 x 0.1 d)

$$\text{rate} = k[\text{H}^+][\text{CH}_3\text{COCH}_3] \\ k = \frac{\text{rate}}{[\text{H}^+][\text{CH}_3\text{COCH}_3]} =$$

$$\frac{9 \times 10^{-5}}{3.0 \times 10^{-5}} = 3.0$$

mol<sup>-1</sup> dm<sup>3</sup> s<sup>-1</sup>

[H<sup>+</sup>][CH<sub>3</sub>COCH<sub>3</sub>]

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