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6 ...

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centimeter (cm) of water at 4 ° C and 760 mm Hg
Mass = quantity of material Weight = mass x
gravitational force Length Meter (m) = 100 cm = 1000
millimeters (mm) = 39.37 in. Volume Liter (L) = volume
of 1 kilogram (kg) of H. 2.

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Mixture and Compound Types of Reactions and
Changes Separation of Mixture Melting Point of Ice and
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Dissolve the samples in 100mL of hot deionized water and add 2 drops of phenolphthalein. Use the pH meter to record the initial pH of the solution. Add more than sufficient amounts of NaOH to the vitamin C samples (~10mL excess) and titrate the excess NaOH with the 0.1M HCl standard solution using a pH meter.

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Determine the mass of the cylinder using the centigram balance. Calculate the density of the metallic cylinder. (Recall that $d=m/V$.) Calculate the Percent Error in your density determination knowing that the theoretical values of the cylinder densities are:

Brass:.....8.50 g/cm³.

Copper:.....8.96 g/cm³.

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