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Writing Ionic Formulas:

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Naming Ionic and Molecular

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~~Basic Introduction~~ How To
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Method for Writing Chemical
Formulas* ~~How to Name
Chemicals Made Easy~~ Naming
~~Ionic Compounds with
Transition Metals~~
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Ions* How to Memorize The
Polyatomic Ions - Formulas,
Charges, Naming - Chemistry
*Naming Compounds in
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Introduction

Pearson Chemistry Chapter
11: Section 1: Describing
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Formulas

How to Write Chemical
Formulas from Compound Names
Pearson Chemistry: Chapter
9: Section 5: The Laws
Governing How Compounds are
Formed *Naming Ionic
Compounds (HD)*

Lewis Diagrams Made Easy:
How to Draw Lewis Dot
Structures ~~Common name,~~
~~chemical name and formula of~~
~~some chemical compounds..~~
Valence Electrons and the
Periodic Table *Naming Ionic
Compounds Chemistry Lesson:
Names \u0026 Formulas of*

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~~Molecular Compounds Pearson
Chemistry Chapter 9: Section
3: Naming and Writing~~

~~Formulas for Molecular
Compounds GCSE Science
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Formula" Balancing Chemical
Equations Practice Problems~~

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1: Naming Ions Empirical
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Formula Introduction~~

~~Covalent Compounds: Writing
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Chapter 9: Chemical Names

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Monatomic Ion. Consists of a single atom with a positive or negative charge resulting from the loss of gain of one or more valence electrons and behaves as a unit, yet it still has a charge. (+ or -). Determined by the amount of electrons lost or gained.

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To name a binary molecular compound, use the following guidelines: 1. Write the names of the elements in the order listed in the formula 2. Use prefixes appropriately to indicate

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Chemical Names Formulas

the number of each kind of atom 3. End the name of the second element with the suffix -ide

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First, confirm that the compound is a binary molecular compound. Name the elements in the order listed in the formula. Use prefixes to indicate the number of each kind of atom. Omit the prefix -mono when the formula contains only one atom of the first element in the name. The suffix of the name of the second element is -ide.

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1. when the name of the anion (X) ends in -ide the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid) 2. when the anion name ends in -ite the acid name is the stem of the anion with the suffix -ous followed by the word acid (ex. H₂SO₃ -- X=sulfite = sulfurous acid)

~~CHAPTER 9 Chemical Names And
Formulas Test Review ...~~

Writing the Names and
Formulas for Bases. A base

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is a compound that produces OH^- in water. When naming a base, you name it like any other compound that starts with a regular or transition metal. Ex: NaOH = sodium hydroxide. When writing the formula for a base, remember to balance charges. Ex: magnesium hydroxide = $\text{Mg}(\text{OH})_2$

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Chapter 9 Chemical Names and
Formulas⁸³ SECTION 9.3

NAMING AND WRITING FORMULAS
FOR MOLECULAR COMPOUNDS

(pages 268-270) This section
explains the rules for
naming and writing formulas
for binary molecular
compounds. Naming Binary
Molecular Compounds (pages
268-269) 1. Circle the
letter of the type(s) of
elements that form binary
molecular compounds.

~~Name Date Class CHEMICAL
NAMES AND FORMULAS 9~~

SECTION 9.2 NAMING AND
WRITING FORMULAS FOR IONIC

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Section Review Answer Key
COMPOUNDS 1. Write the formulas for these binary ionic compounds. c. potassium iodide 2. write the formulas for the compounds formed c. b. f. 3. Name the following binary ionic compounds. sodium sulfide g. CuCl: h. snc.l. a. MnO₂ c. d. SrBr₂ 221
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Chemical Names and Formulas
281 CHAPTER 9 Assessment 42.
a. 2- b. 1+ c. 1- d. 3+ 43.
a. 2+ b. 2+ c. 3+ d. 1+ 44.
a. barium ion b. iodide ion
c. silver ion d. mercury(II)
ion 45. cyanide, CN^- and
hydroxide, OH^- 46. a.
hydroxide ion b. lead(IV)
ion c. sulfate ion d. oxide
ion 47. zero; A compound is
electrically neutral. 48.
The symbols for the cation
and

~~CHAPTER 9 Study Guide — Quia~~
Chapter 9 Chemical Names
Formulas Chapter 9 -
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Match. Gravity. Created by.
bibiloni1901. Terms in this
set (20) Monatomic Ion.
Consists of a single atom
with a positive or negative
charge resulting from the
loss of gain of one or more
valence

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and Formulas" Tools. Copy
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activities; Start over;
Help; Use these activities
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chapter. A B; anion: any
atom or group of atoms that
has a negative charge:

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Chapter 9 Practice Test -
Naming and Writing Chemical
Formulas Answer Section

MATCHING 1. ANS: A PTS: 1

DIF: L1 REF: p. 253 OBJ:

9.1.1 Identify the charges
of monatomic ions by using
the periodic table, and name
the ions. STA: Ch.5.a 2.ANS:
H PTS: 1 DIF: L1 REF: p. 254

~~Chapter 9 Practice Test~~

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~~Naming and Writing Chemical Formulas~~

Chemistry (12th Edition)
answers to Chapter 9 -
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- 9.1 Naming Ions - 9.1
Lesson Check - Page 269 4
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