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SECTION 13-2 SHORT ANSWER

Answer the following questions in the  
space provided. 1. Following are  
statements about the dissolving  
process. Explain each one at the  
molecular level. a. Increasing the  
pressure of a solute gas above a liquid  
solution increases the solubility of the  
gas in the liquid. b.

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2 Answers

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### CHAPTER 13 AUDITING THE INVENTORY MANAGEMENT PROCESS

Answers to Review Questions 13-2

The inventory process is affected by control activities in the revenue, purchasing, and human resource management processes. The purchasing process controls the acquisition and payment of raw materials and overhead costs. The cost of direct and indirect labor assigned to inventory is controlled through the ...

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Chapter 13 221 Section 13.4

Solutions of Gases in Liquids Goals To describe the process by which gases dissolve in liquids. To explain why gases have a solubility limit in liquids. To explain why increased partial pressure of a gas over a liquid will lead to an increase in the solubility of that gas.

Chapter 13 Solution Dynamics

Chapter 13

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite



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soluble in water (a) HCl c an ionic compound that is not very soluble in water (b)  $\text{NaNO}_3$

13 Ions in Aqueous Solutions and Colligative Properties

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CHAPTER . 12 . REVIEW "-- Solutions

. SHORT ANSWER Answer the

following questions in the space

provided. 1. Describe the errors made

by the following students in making

molar solutions. 3. James needs a

0.600 M solution of KCl. He measures

out 0.600 g of KCl and adds 1 L of

water to the solid. James made several

errors.

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12 SECTION 1 SHORT ANSWER 1. c 2.

a 3. b 2. a. alcohol b. water c. the gels

3. The mixture is a colloid. The

properties are consistent with those

reported in Table 3 on page 404 of

the text. The particle size is small, but

not too small, and the mixture

CHAPTER 12 REVIEW Solutions -

Weebly

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Estimate 30 balls lost per game. 81

games  $\times 310 \text{ g balls} = 34.$

Estimate 14 lb per ...

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Goods and Services; 3.3 Changes in  
Equilibrium Price and Quantity: The  
Four-Step Process; 3.4 Price Ceilings  
and Price Floors; 3.5 Demand, Supply,  
and Efficiency; Key Terms; Key  
Concepts and Summary; Self-Check  
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Chapter 13 Review, Section 1: Ions in  
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that are studying ionic solutions.

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