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microwaves. Key. (2) Sol.: P 4000 current  
leads the voltage by  $30^\circ$ . The power  $20 \text{ E } 10$   
 $hc \text{ hc } 10 \text{ hc } 17 \text{ } 4000 \text{ } 4 \text{ } 3 \text{ } 10 \text{ } 6.6 \text{ } 108 \text{ } 34 \text{ } 17 \text{ } 4 \text{ } 9$   
 $19.8 \text{ } 10 \text{ } 4 \text{ } 4.9 \text{ } 10 \text{ } 9 \text{ } 49 \text{ } 10 \text{ } 10 \text{ } 49\text{A} \text{ } 0.1\text{A} \text{ } 100\text{A} \text{ } 0$   
it is X-ray (2) is correct. 16.

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derived wrt slow

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solu.*

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is passed through a cation exchanger. The  
chloride ions obtained in solution were  
treated with excess of  $\text{AgNO}_3$  to give 4.78 g  
of  $\text{AgCl}$  (molar mass = 143.5 g mol<sup>-1</sup>).

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an ideal solution. At 373 K, the vapour pressures of the two liquid components (heptane and octane) are 105 kPa and 45 kPa respectively. Vapour pressure of the solution obtained by mixing 25.0g of heptane and 35 g of octane will be (molar mass of heptane = 100 g mol<sup>-1</sup> and of octane = 114 g mol<sup>-1</sup>)

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